

Visible Light-Mediated DNA Damage

Amelia C. McCue and Thomas A. Shell

Department of Chemistry
Saint Anselm College, Manchester NH

Introduction

Radical cleavage of DNA and RNA has been proven useful in studying the structures of these biological complexes

Light-dependent radical cleavage is desirable because initiation and termination of radical production can be controlled

Compounds used for generating radicals in a light-mediated fashion require absorption of relatively high energy UV light

UV light is absorbed by DNA bases causing damage

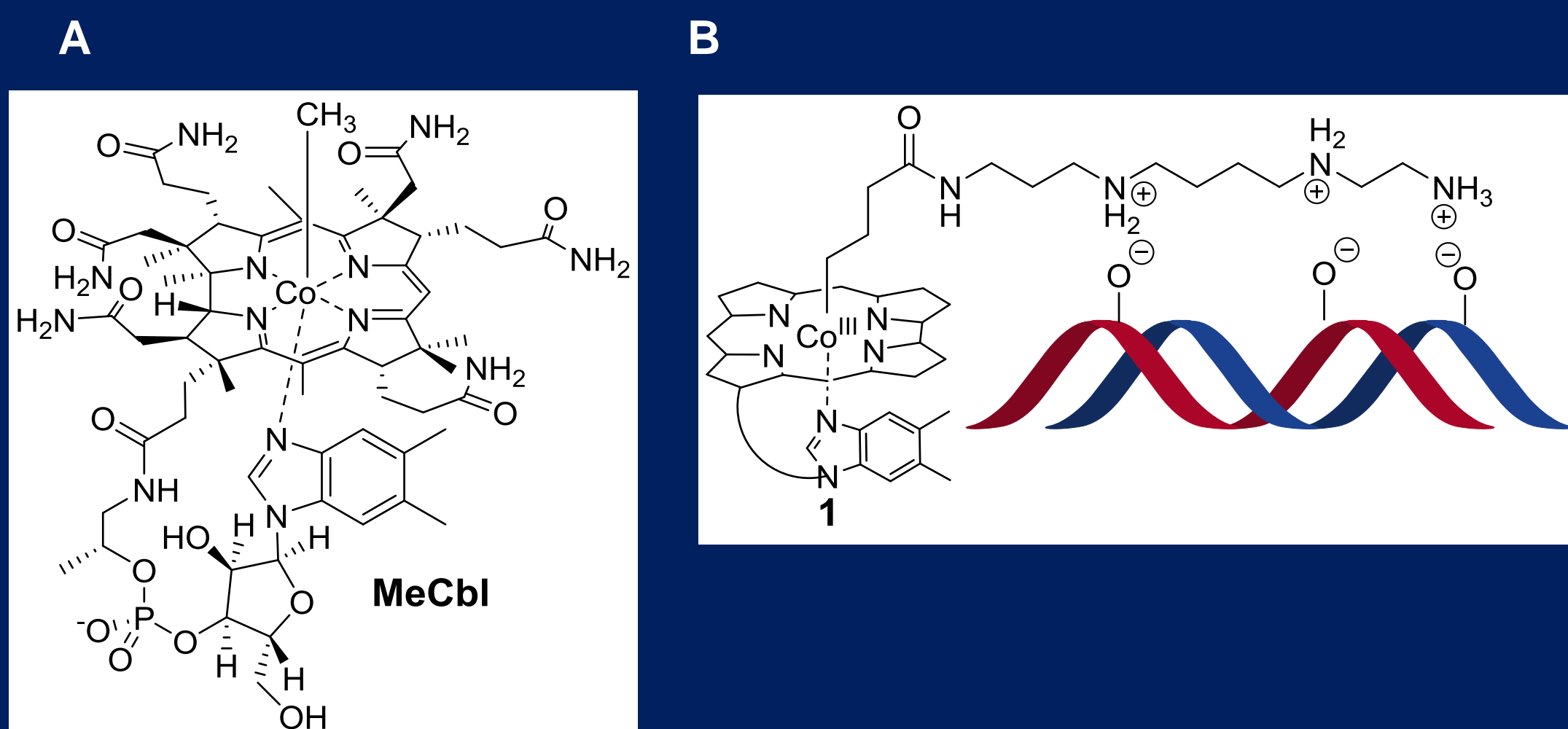
Methylcobalamin and other alkylcobalamins are known to undergo photolysis to generate methyl radicals upon exposure to green light (530 – 580 nm)

Green light is not absorbed by DNA bases

We anticipated that methylcobalamin in combination with green light would cause DNA cleavage without damage due to the light

Light-dependent DNA Damage by Methylcobalamin

Scheme 1. A) Methylcobalamin; (B) Cartoon of spermine-alkylcobalamin conjugate (**1**) binding to DNA; C) Light-mediated DNA cleavage by methylcobalamin.



Visible Light-Mediated Cleavage of pBR322 DNA by Methylcobalamin (phosphate buffer, pH 7.5)

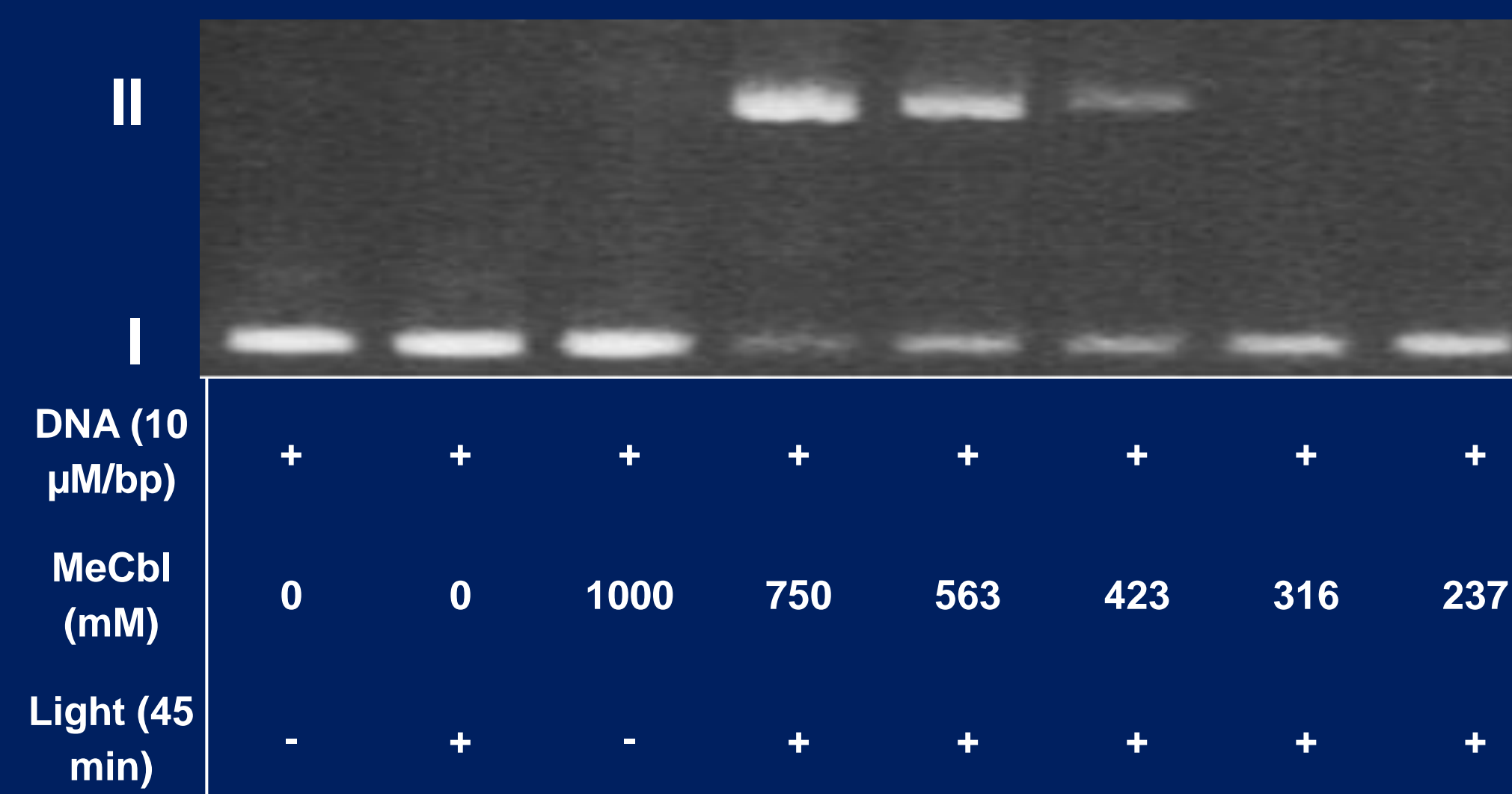


Fig. 1. Light-mediated DNA cleavage dependence on MeCbl concentration.

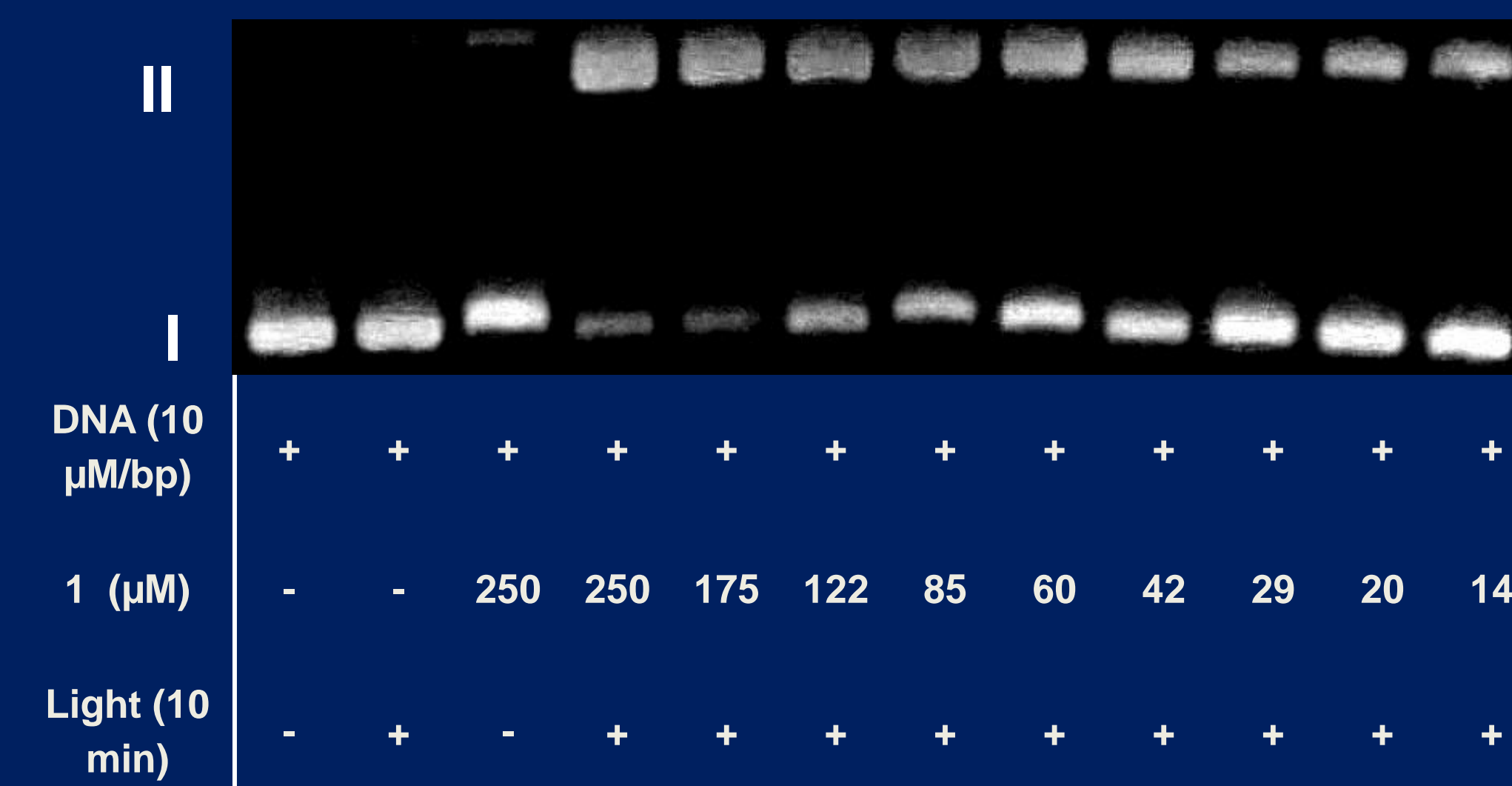


Fig. 3. DNA cleavage dependence on the concentration of spermine-alkylcobalamin compound (**1**).

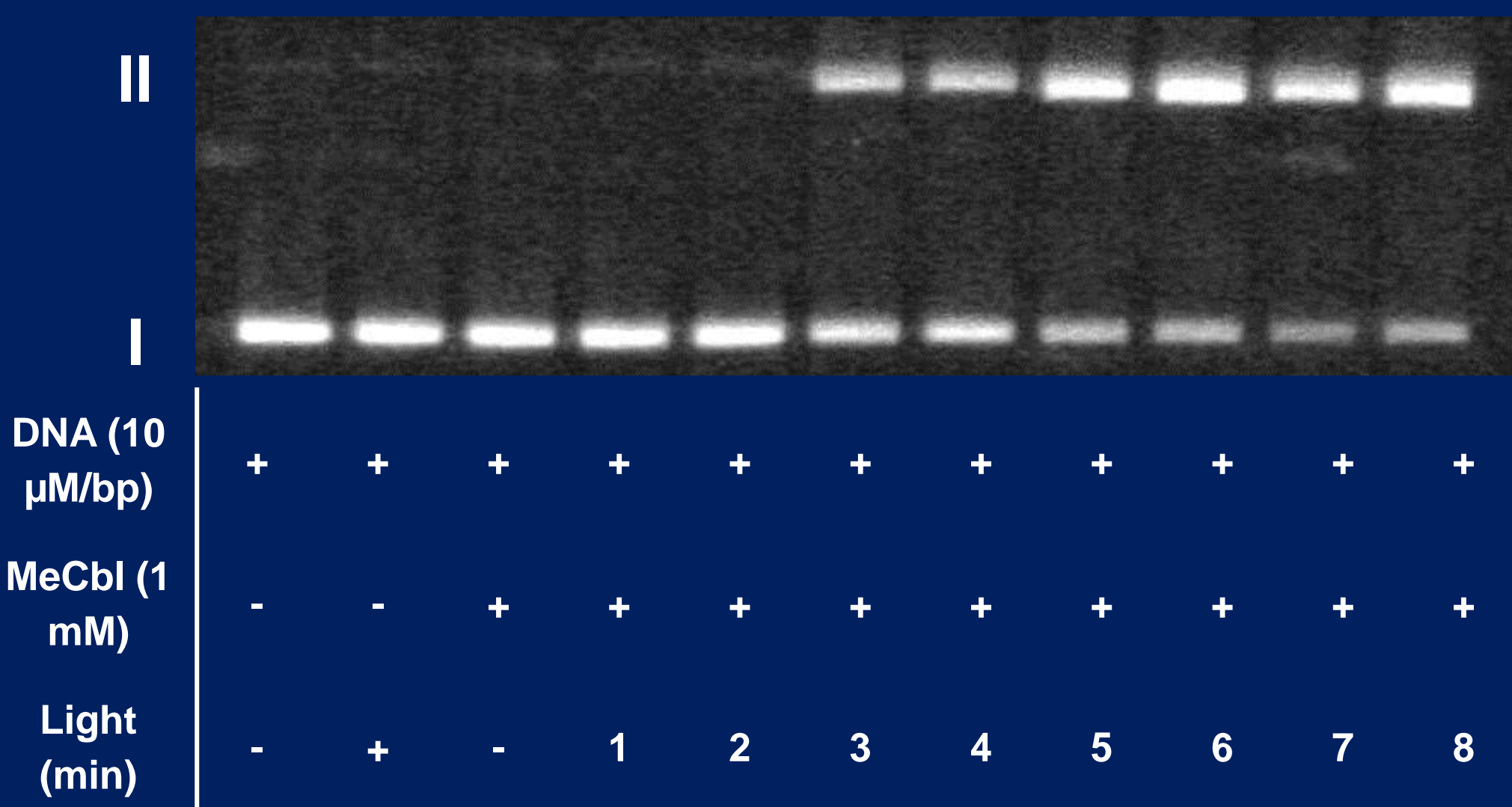


Fig. 2. DNA cleavage by MeCbl dependence on light exposure.

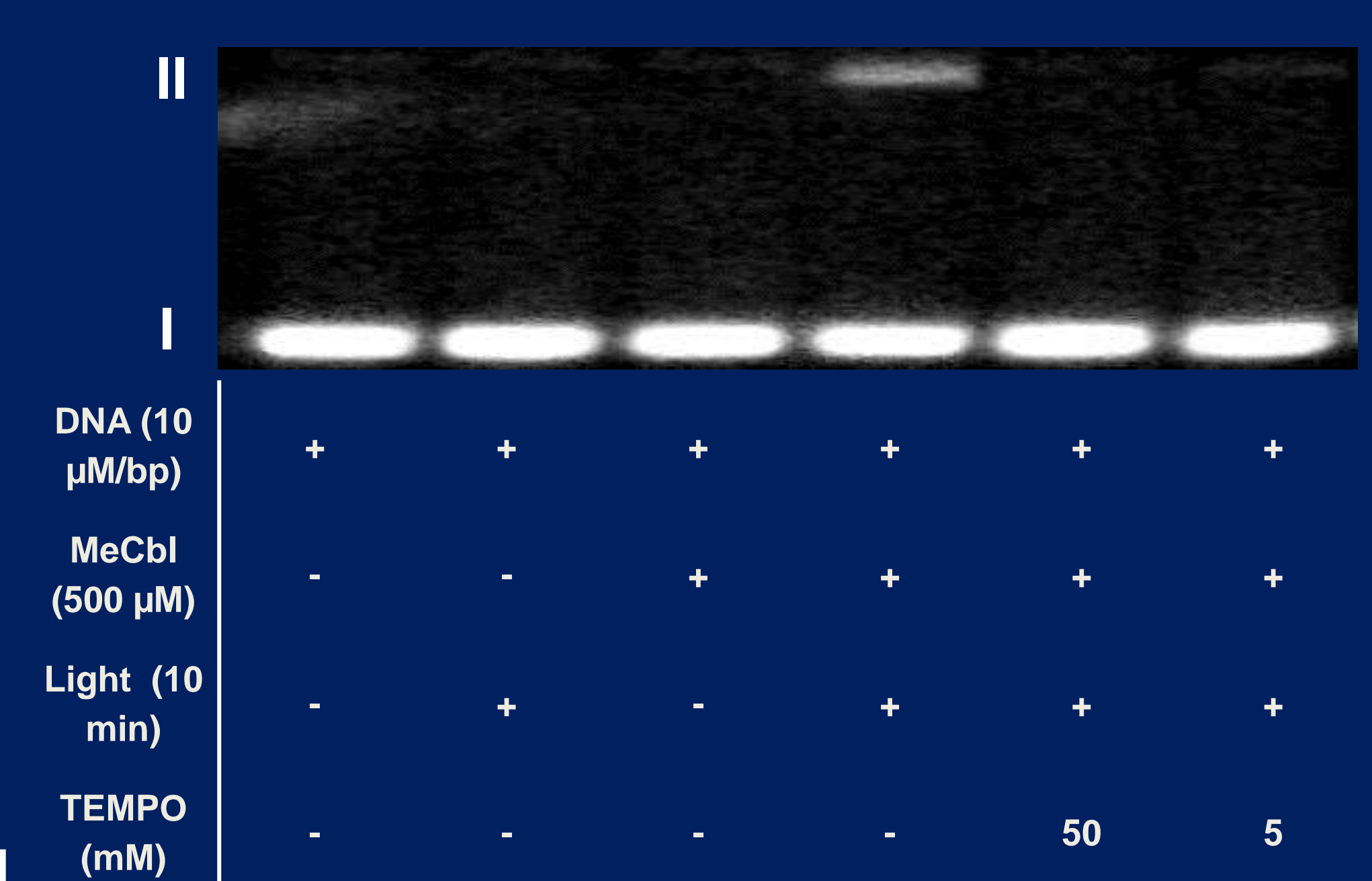


Fig. 4. DNA cleavage inhibition by TEMPO, a radical scavenger.

Conclusions

Our results show that methylcobalamin, an alkylcobalamin, cleaves DNA in a green light-mediated manner

DNA strand scission results from the generation of methyl radicals

Green light does not damage DNA; in contrast to UV light

Conjugation to a DNA binding agent, spermine, improved the ability alkylcobalamins to cause DNA damage

We anticipate that this could be useful for researchers studying DNA and RNA structure

Acknowledgements

Funding for this research was provided by NH-INBRE. I would like to thank Dr. Shell for his guidance and support as research advisor.